

# Rankine Cycle



## Reading

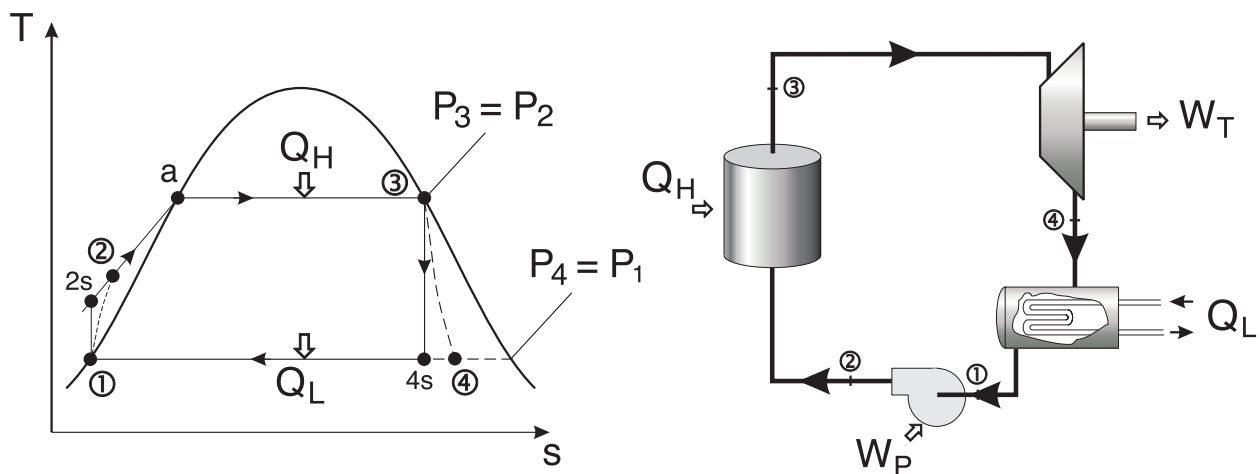
9-2 → 9-7

## Problems

9-16, 9-23, 9-30, 9-38, 9-43, 9-84

## Definitions

- working fluid is alternately vaporized and condensed as it recirculates in a closed cycle
- water is typically used as the working fluid because of its low cost and relatively large value of enthalpy of vaporization
- the standard vapour cycle that excludes internal irreversibilities is called the **Ideal Rankine Cycle**



- the condensation process is allowed to proceed to completion between state points 4 → 1
  - provides a saturated liquid at 1
- the water at state point 1 can be conveniently pumped to the boiler pressure at state point 2
- but the water is not at the saturation temperature corresponding to the boiler pressure
- heat must be added to change the water at 2 to saturated water at 'a'
- when heat is added at non-constant temperature (between 2 — a), the cycle efficiency will decrease

## Analyze the Process

Assume steady flow,  $KE = PE = 0$ .

From a 1st law balance, we know

$$\text{energy in} = \text{energy out}$$

Device	1st Law Balance
<b>Boiler</b>	$h_2 + q_H = h_3 \Rightarrow q_H = h_3 - h_2$ (in)
<b>Turbine</b>	$h_3 = h_4 + w_T \Rightarrow w_T = h_3 - h_4$ (out)
<b>Condenser</b>	$h_4 = h_1 + q_L \Rightarrow q_L = h_4 - h_1$ (out)
<b>Pump</b>	$h_1 + w_P = h_2 \Rightarrow w_P = h_2 - h_1$ (in)

The net work output is given as

$$\begin{aligned}w_T - w_P &= (h_3 - h_4) - (h_2 - h_1) \\&= (h_3 - h_4) + (h_1 - h_2)\end{aligned}$$

The net heat supplied to the boiler is

$$q_H = (h_3 - h_2)$$

The Rankine efficiency is

$$\begin{aligned}\eta_R &= \frac{\text{net work output}}{\text{heat supplied to the boiler}} \\&= \frac{(h_3 - h_4) + (h_1 - h_2)}{(h_3 - h_2)}\end{aligned}$$

If we consider the fluid to be incompressible

$$(h_2 - h_1) = v(P_2 - P_1)$$

Since the actual process is irreversible, an isentropic efficiency can be defined such that

$$\text{Expansion process} \Rightarrow \text{Isentropic efficiency} = \frac{\text{actual work}}{\text{isentropic work}}$$

$$\text{Compression process} \Rightarrow \text{Isentropic efficiency} = \frac{\text{isentropic work}}{\text{actual work}}$$

Both isentropic efficiencies will have a numerical value between 0 and 1.

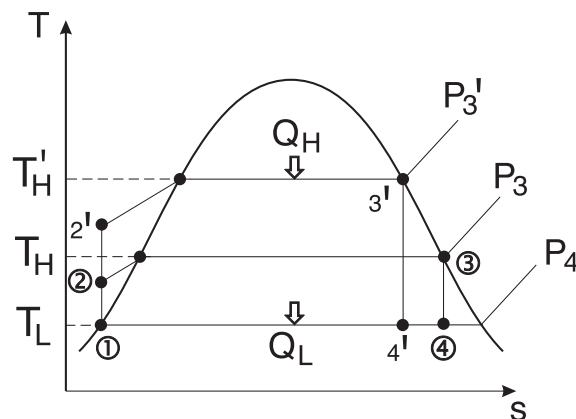
## Effects of Boiler and Condenser Pressure

We know the efficiency is proportional to

$$\eta \propto 1 - \frac{T_L}{T_H}$$

The question is  $\rightarrow$  how do we increase efficiency  $\Rightarrow T_L \downarrow$  and/or  $T_H \uparrow$ .

### 1. INCREASED BOILER PRESSURE:



- an increase in boiler pressure results in a higher  $T_H$  for the same  $T_L$ , therefore  $\eta \uparrow$ .
- but 4' has a lower quality than 4
  - wetter steam at the turbine exhaust

- results in cavitation of the turbine blades
- $\eta \downarrow$  plus  $\uparrow$  maintenance
- quality should be  $> 90\%$  at the turbine exhaust

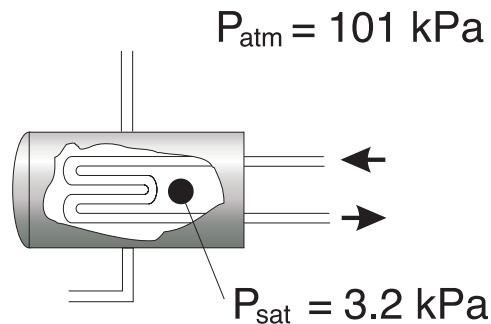
## 2. LOWER $T_L$ :

- we are generally limited by the **TER** (lake, river, etc.)

eg. lake @  $15^\circ\text{C} + \underbrace{\Delta T = 10^\circ\text{C}}_{\text{resistance to HT}} = 25^\circ\text{C}$

$\Rightarrow P_{\text{sat}} = 3.2 \text{ kPa}.$

- this is why we have a condenser
  - the pressure at the exit of the turbine can be less than atmospheric pressure
  - the closed loop of the condenser allows us to use treated water on the cycle side
  - but if the pressure is less than atmospheric pressure, air can leak into the condenser, preventing condensation



## 3. INCREASED $T_H$ BY ADDING SUPERHEAT:

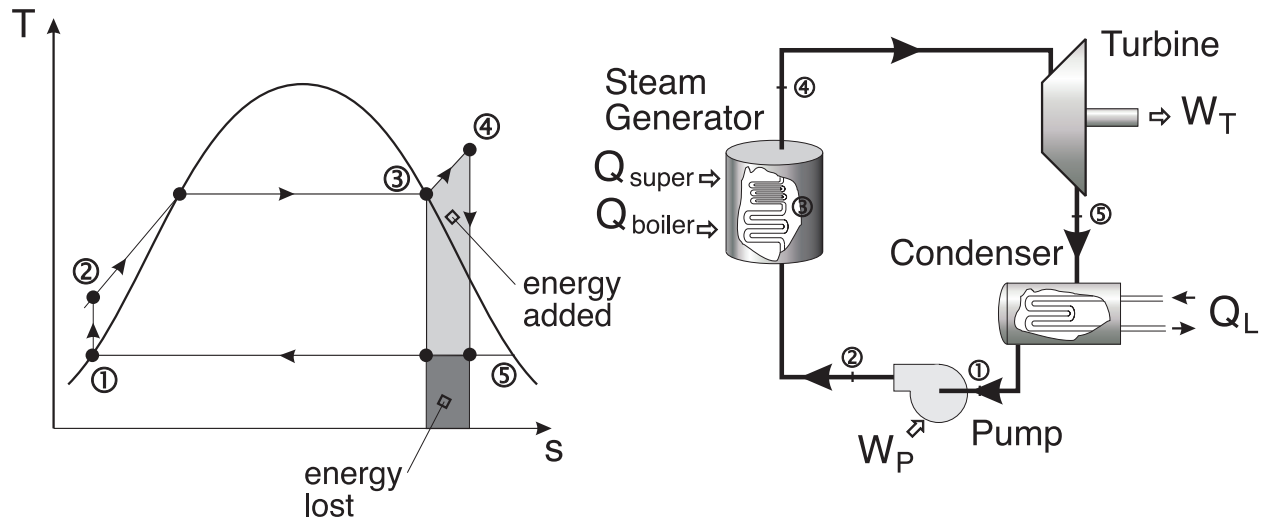
- the average temperature at which heat is supplied in the boiler can be increased by superheating the steam
  - dry saturated steam from the boiler is passed through a second bank of smaller bore tubes within the boiler until the steam reaches the required temperature

The advantage is

$$\eta = \frac{W_{\text{net}} \uparrow}{Q_H \uparrow} \quad \text{overall } \uparrow$$

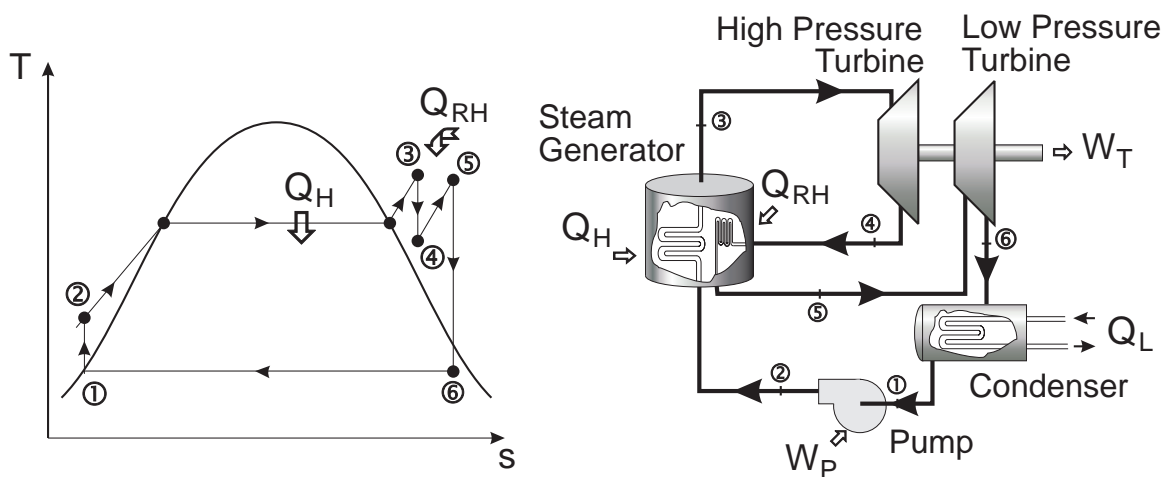
The value of  $\bar{T}_H$ , the mean temperature at which heat is added, increases, while  $\bar{T}_L$  remains constant. Therefore the efficiency increases.

- the quality of the turbine exhaust increases, hopefully where  $x > 0.9$
- with sufficient superheating the turbine exhaust can fall in the superheated region.



## Rankine Cycle with Reheat

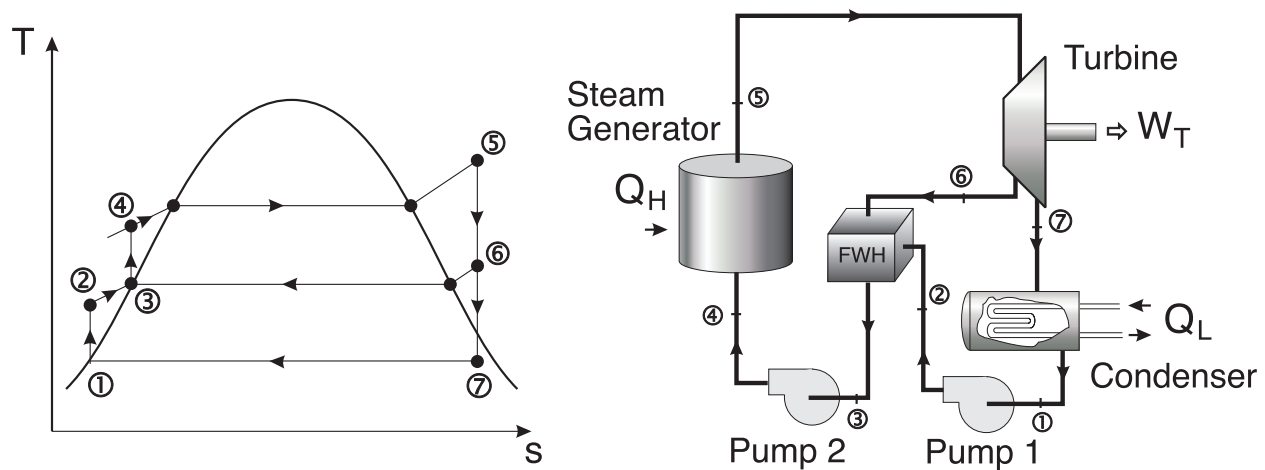
- the wetness at the exhaust of the turbine should be no greater than 10% - this can result in physical erosion of the turbine blades
- but high boiler pressures are required for high efficiency - tends to lead to a high wetness ratio
- to improve the exhaust steam conditions, the steam can be reheated with the expansion carried out in two steps



- the temperature of the steam entering the turbine is limited by metallurgical constraints
- modern boilers can handle up to  $30 \text{ MPa}$  and a maximum temperature of  $T_{\text{max}} \approx 650^\circ\text{C}$ .
- newer materials, such as ceramic blades can handle temperatures up to  $750^\circ\text{C}$ .

## Rankine Cycle with Regeneration

- Carnot cycle has efficiency:  $\eta = 1 - T_L/T_H$ 
  - add  $Q_H$  at as high a  $T_H$  as possible
  - reject  $Q_L$  at as low a  $T_L$  as possible
- the Rankine cycle can be used with a **Feedwater Heater** to heat the high pressure sub-cooled water at the pump exit to the saturation temperature
  - most of the heat addition ( $Q_H$ ) is done at high temperature



## Feedwater Heaters

There are two different types of feedwater heaters

1. **OPEN FWH:** the streams mix  $\rightarrow$  high temperature steam with low temperature water at constant pressure
2. **CLOSED FWH:** a heat exchanger is used to transfer heat between the two streams but the streams do *not* mix. The two streams can be maintained at different pressures.

## 1. *OPEN FWH*:

- working fluid passes isentropically through the turbine stages and pumps
- steam enters the first stage turbine at state 1 and expands to state 2 - where a fraction of the total flow is bled off into an open feedwater heater at  $P_2$
- the rest of the steam expands into the second stage turbine at state point 3 - this portion of the fluid is condensed and pumped as a saturated liquid to the FWH at  $P_2$
- a single mixed stream exists the FWH at state point 6

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Analysis:

- we must determine the mass flow rates through each of the components.  
By performing an mass balance over the turbine

$$\dot{m}_2 + \dot{m}_3 = \dot{m}_1 \quad (1)$$

If we normalize Eq. (1) with respect the total mass flow rate  $\dot{m}_1$

$$\frac{\dot{m}_2}{\dot{m}_1} + \frac{\dot{m}_3}{\dot{m}_1} = 1 \quad (2)$$

Let the flow at state point 2 be

$$y = \frac{\dot{m}_2}{\dot{m}_1}$$

Therefore

$$\frac{\dot{m}_3}{\dot{m}_1} = 1 - y \quad (3)$$

Assuming no heat loss at the FWH, establish an energy balance across the FWH

$$yh_2 + (1 - y)h_5 = 1 \cdot h_6$$

$$y = \frac{h_6 - h_5}{h_2 - h_5} = \frac{\dot{m}_2}{\dot{m}_1}$$

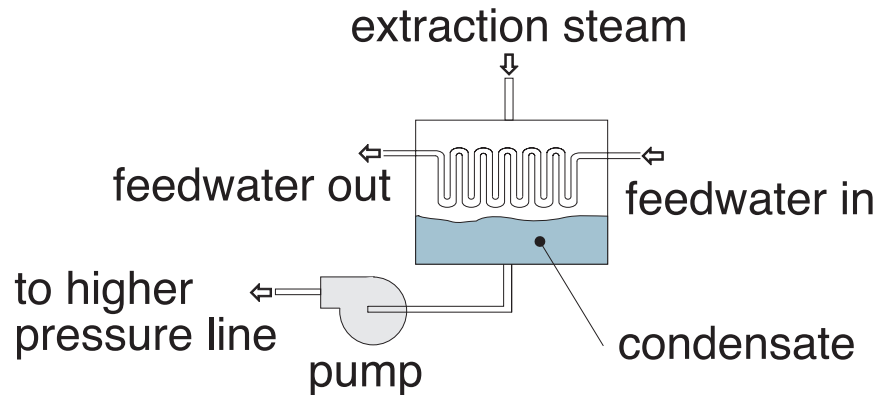
and

$$1 - y = \frac{\dot{m}_3}{\dot{m}_1}$$

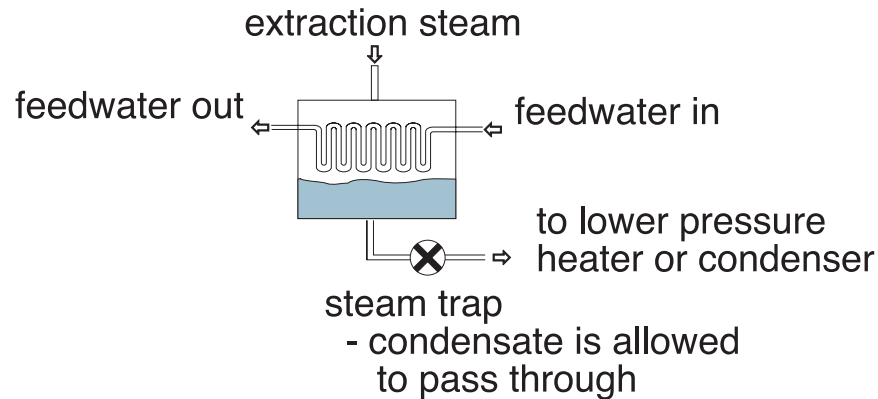
## 2. *CLOSED FWH*:

- two variations exist

(a) pump the condensate back to the high pressure line



- (b) – a steam trap is inserted in the condensed steam line that allows only liquid to pass
- liquid is passed to a low pressure region such as the condenser or a low pressure heater

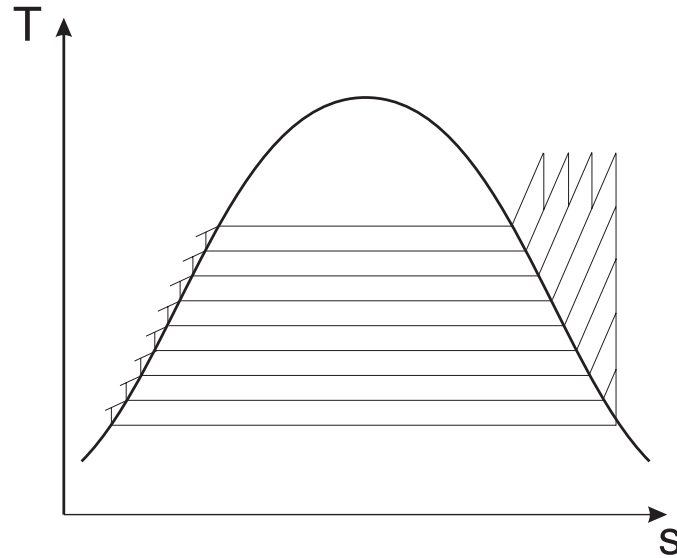


- the incoming feedwater does not mix with the extracted steam
  - both streams flow separately through the heater
  - the two streams can have different pressures



## Other Topics

### *“IDEAL” RANKINE CYCLE:*



- too expensive to build
- requires multiple reheat and regeneration cycles
- approaches Carnot efficiency

### *TOPPING CYCLE (BINARY CYCLE):*

- involves two Rankine cycles running in tandem with different working fluids such as mercury and water
- why:
  - typically a boiler will supply energy at  $1300 - 1400\text{ }^{\circ}\text{C}$
  - but  $T_{critical}$  for water is  $374.14\text{ }^{\circ}\text{C}$ 
    - \* most energy is absorbed below this temperature
    - \* high  $\Delta T$  between the boiler source and the water leads to a major source of irreversibilities
  - $T_{critical}$  for mercury is about  $1500\text{ }^{\circ}\text{C}$ 
    - \* no need for superheating
  - combine the large enthalpy of evaporation of water at low temperatures with the advantages of mercury at high temperatures
  - in addition, the mercury dome leads to a high quality at the exit of the turbine