

**Week 5: Lecture 1****Refrigeration Cycle**

A refrigeration system removes thermal energy from a low-temperature region and transfers heat to a high-temperature region.

The Carnot cycle can serve as the initial model of the ideal refrigeration cycle.

- operates as a reversed heat cycle engine
- transfer a quantity of heat,  $Q_L$  from a cold source at temperature,  $T_L$

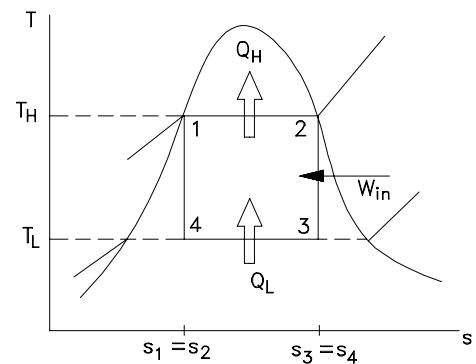
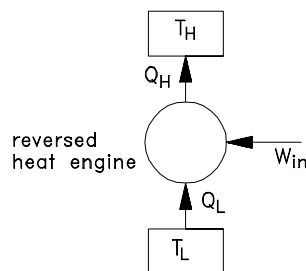
where:

$$Q_L = T_L(s_3 - s_2)$$

$$Q_H = T_H(s_4 - s_1)$$

$$W_{in} = Q_{net}$$

$$= (T_H - T_L)(s_3 - s_2)$$

**Coefficient of Performance**

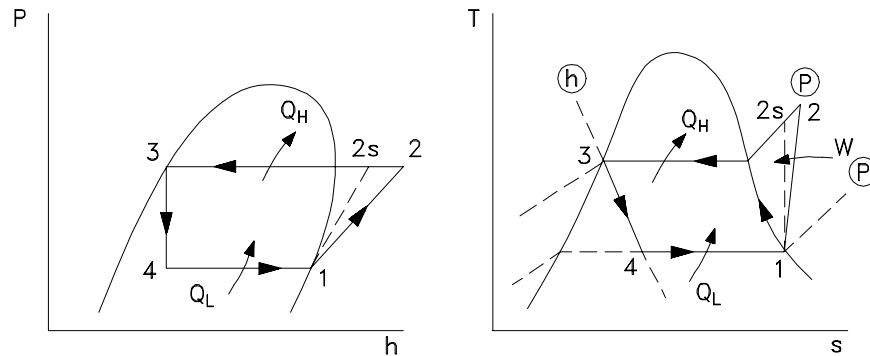
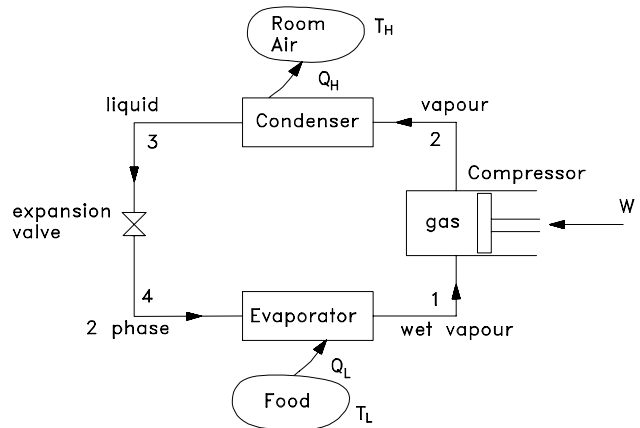
$$COP = \frac{\text{benefit}}{\text{cost}}$$

$$COP_{refrig} = \frac{Q_L}{W_{in}} = \frac{1}{\frac{T_H}{T_L} - 1} = \frac{T_L}{T_H - T_L}$$

$$COP_{heat\ pump} = \frac{Q_H}{W_{in}} = \frac{1}{1 - \frac{T_L}{T_H}} = \frac{T_H}{T_H - T_L}$$

**Week 5: Lecture 1****Vapour Compression Refrigeration Cycle**Assumptions for Ideal VCRC

- irreversibilities within the evaporator, condenser and compressor are ignored
- no frictional pressure drops
- refrigerant flows at constant pressure through the two heat exchangers
- stray heat losses to the surroundings are ignored
- compression process is isentropic



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- 1-2s: isentropic compression of the refrigerant
  - 2s-3: heat transfer from the refrigerant as it flows at constant pressure through the condenser
  - 3-4: throttling process from state 3 to a two phase liquid-vapour mixture at state 4
  - 4-1: heat transfer to the refrigerant as it flows at constant pressure through the evaporator
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