

Week 5: Lecture 1

Refrigeration Cycle

A refrigeration system removes thermal energy from a low-temperature region and transfers heat to a high-temperature region.

The Carnot cycle can serve as the initial model of the ideal refrigeration cycle.

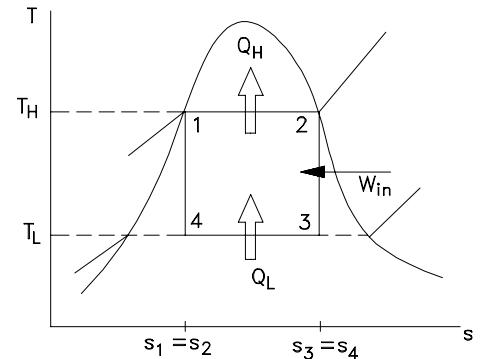
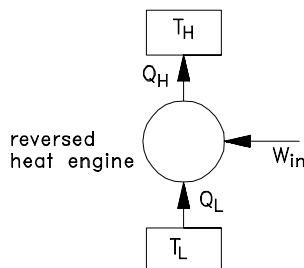
- operates as a reversed heat cycle engine
- transfer a quantity of heat, Q_L from a cold source at temperature, T_L

where:

$$Q_L = T_L(s_3 - s_2)$$

$$Q_H = T_H(s_4 - s_1)$$

$$W_{in} = Q_{net}$$



Coefficient of Performance

$$COP = \frac{\text{benefit}}{\text{cost}}$$

$$COP_{refrig} = \frac{Q_L}{W_{in}} = \frac{1}{\frac{T_H}{T_L} - 1} = \frac{T_L}{T_H - T_L}$$

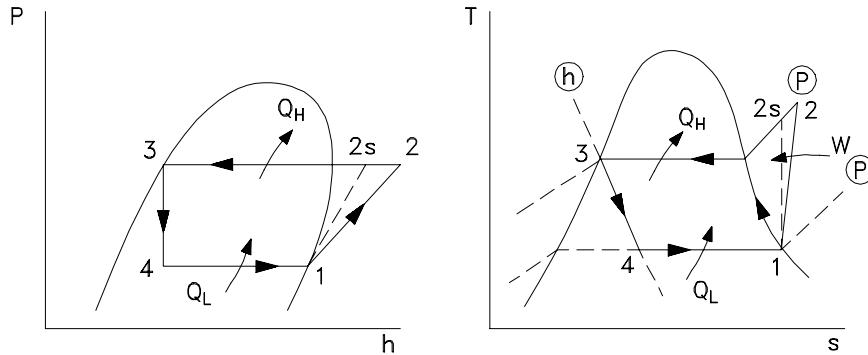
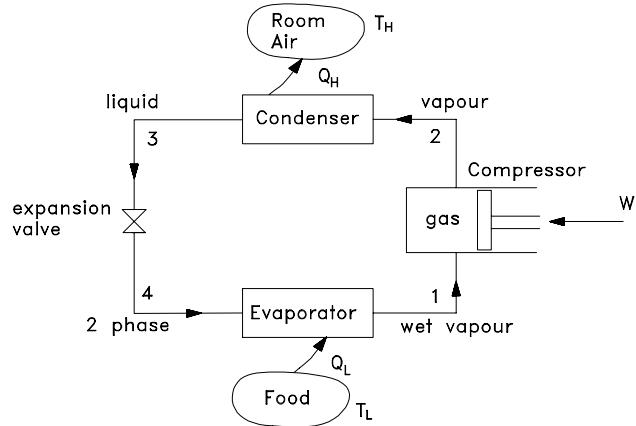
$$COP_{heat\ pump} = \frac{Q_H}{W_{in}} = \frac{1}{1 - \frac{T_L}{T_H}} = \frac{T_H}{T_H - T_L}$$

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Vapour Compression Refrigeration Cycle

Assumptions for Ideal VCRC

- irreversibilities within the evaporator, condenser and compressor are ignored
- no frictional pressure drops
- refrigerant flows at constant pressure through the two heat exchangers
- stray heat losses to the surroundings are ignored
- compression process is isentropic



1-2s: isentropic compression of the refrigerant
 2s-3: heat transfer from the refrigerant as it flows at constant pressure through the condenser
 3-4: throttling process from state 3 to a two phase liquid-vapour mixture at state 4
 4-1: heat transfer to the refrigerant as it flows at constant pressure through the evaporator
